H₃⁺ in the Diffuse Interstellar Medium The Problem of the Ionization

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Diffuse clouds

Properties:

- Density: $n_H \approx 100 \text{ cm}^{-3}$
- embedded in the ISRF
- Kinetic temperature $T_{kin} \approx T_{01} \approx 70 \text{ K}$
- Transition between atomic and molecular hydrogen

Observation in absorption possible

- Far UV : H, H₂, HD, CO, ...
- Visible: OH, CH, CH⁺, CN,
 - C₂, C₃, ...
- IR: H₃⁺
- Radio : HCO⁺, HOC⁺, NH₃, HCN, HNC, H₂S, ...





Interest to study diffuse clouds

Simple chemistry → good place to understand the physics of ISM

Two fundamental questions:

dissipation of energy

- formation process of CH⁺
- rotationnal excitation of H₂

ionization

• the formation of many molecules is initiated by cosmic rays ionization H_3^+ , HD, OH, ...

Formation:	$H_2 + cosm. ray \longrightarrow H_2^+ + e^-$ $H_2^+ + H_2 \longrightarrow H_3^+ + H_3$	k = 0.96 × ζ k = 2.1 10 ⁻⁹
Destruction:	H ₃ ⁺ + e ⁻ ▶	k = 6.80×10 ⁻⁸ (T/300) ^{-0.5}

n(HD), n(OH), n(H₃⁺) ∝ ζ

Determination of the ionization rate by cosmic rays

(Black et Dalgarno 1973, Black et al. 1978, Federman et al. 1996, Le Petit et al. 2001)



H₃⁺ in diffuse clouds

Observation:

- detected on 8 diffuse lines of sight $N(H_3^+) / E(B-V) \sim \text{some } 10^{14}$		E(B-V)	N(H ₃ ⁺)
=> 10 × higher than dense clouds	Cyg. OB2 12	3.35	2.02 (14)
- in diffuse medium near the Galactic center	Cyg. OB2 5	1.99	~ 3 (14)
(Oka et al. 2005)	HD 183143	1.28	~ 2 (14)
	HD 20041	0.70	1.74 (14)
Model: n _H = 100 cm ⁻³ χ = 1	WR 104	2.10	~ 2 (14)
T = 60 K ζ = 5 ×10 ¹⁷ s ⁻¹	WR 118	4.13	~ 4 (14)
$N_{\rm H} = 10^{21} {\rm cm}^{-2}$	WR 121	1.68	1.12 (14)
\downarrow	ζPer	0.32	8.0 (13)
$N(H_3^+) = 8 \times 10^{12} \text{ cm}^{-2}$	Gal. center		3.1 (15)
	Ref [.] McCall et al	(2002)	

00Z) McCall et al. (2003) Oka et al. (2005)

The Zeta Per line of sight

A very well studied line of sight

Spectral type : B1 $R_V = 2.8$ (Cardelli et al. 1989) E(B-V) = 0.32 (van Dishoeck & Black 1989)

A very good test for models

	Observation
H ₂	3.2 - 7.1 (20)
H ₂ (J=0)	2.2 - 4.8 (20)
H ₂ (J=1)	1.0 - 2.3 (20)
H ₂ (J=2)	1.1 - 2.4 (18)
H ₂ (J=3)	2.0 - 9.6 (16)
H ₂ (J=4)	1.1 - 2.0 (15)
H ₂ (J=5)	2.3 - 2.8 (14)

	Observ	ations/
Н	5.7(20)	7.1(20)
H_2	3.2(20)	7.1(20)
f	0.53	0.66
T ₀₁	45	75
HD	2.0(15)	1.1(16)
H ₃ +	8.0(13)	
C ⁺	1.8(17)	
С	2.9(15)	3.6(15)
CO	5.4(14)	
CH	1.9(13)	2.0(13)
CH⁺	3.5(12)	
C ₂	1.6(13)	2.2(13)
C ₃	1.0(12)	
CN	2.7(12)	3.3(12)
NH	9.0(11)	
0	0.2(18)	1.0(18)
OH	4.0(13)	
S+	1.7(16)	2.3(16)
S	1.5(13)	2.2(13)
Si⁺	2.8(16)	2.8(14)

Determination of the flux of cosmic rays



 van Dishoeck & Black (1986) all constraints taken into account (at this time ...) models with T and n_H profile

ζ = 4-7×10⁻¹⁷ s⁻¹

• Federman et al. (1996) From OH only : ζ = 1.7×10⁻¹⁷ s⁻¹



Comparison of ζ between different authors not always simple:

H + cosm. ray
$$\rightarrow$$
 H⁺ + e⁻k = 0.46 × ζ (s⁻¹) - Prassad & Huntress (1980)H2 + cosm. ray \rightarrow H2⁺ + e⁻k = 0.96 × ζ (s⁻¹) - A. Dalgarno (priv. com.)H + HH + Hk = 1.50 × ζ (s⁻¹) - A. Dalgarno (priv. com.)H⁺ + H + e⁻k = 0.04 × ζ (s⁻¹) - A. Dalgarno (priv. com.)

The Meudon PDR code (http://aristote.obspm.fr/MIS)



Stationnary model solving:

- Radiative transfer: absorption in the lines of H, H₂, CO, HD, ... absorption in the continuum
- Chemistry: more than 100 chemical species network of more than 1000 chemical reactions photoionization
- Statistical equilibrium of the populations in the levels of H₂, HD, CO, HCO⁺, CS, ... takes into account: radiative and collisional excitation / de-excitations photodissociation
- Thermal balance: heating by photoelectric effect, chemistry, cosmic rays ... cooling in the lines of atoms and molecules

Modelisation of the ζ Per line of sight

(Franck Le Petit, Evelyne Roueff & Eric Herbst, A&A, 2004)

Parameters and hypothesis :

 $R_V = 2.8$ Isothermal model : $T_{01} = 60 \text{ K}$ (45-75 K) $N(H_2) = 4.5 \times 10^{20} \text{ cm}^{-2}$

n_H = 100 cm⁻³ χ= 2

	Relative
	abundance
D/H	1.5 (-5)
O/H	3.2 (-4)
N/H	7.5 (-5)
C/H	1.32 (-4)
S/H	1.86 (-5)
Si/H	2.9 (-5)

Variation of ζ between 1 ×10⁻¹⁷ and 100 ×10⁻¹⁷ s⁻¹ \Leftrightarrow effect on species sensible to ζ

Determination of ζ from H₃⁺



Depends on T

Observations require : $\zeta \sim 100 \times 10^{-17} \text{ cm}^{-3}$

Determination of ζ from OH

n(OH) is highly dependent on T

H⁺ + O \longrightarrow O⁺ + H $k = 6 \times 10^{-10} \text{ e}^{-227/\text{T}} \text{ cm}^3 \text{ s}^{-1}$ 45 K $k = 3.9 \ 10^{-12} \text{ cm}^3 \text{ s}^{-1}$ 75 K $k = 2.9 \ 10^{-11} \text{ cm}^3 \text{ s}^{-1}$





n(HD) $\propto \zeta$ if : 1) It is formed in gas phase by D⁺ + H₂ 2) It is destroyed by photodissociation

after the D/HD transition :

 $HD + H_3^+ \longrightarrow H_2D^+ + H$

n(HD) no more proportionnal to ζ



- still some debates on D/H
- difficult to get a precise value of N(HD)

Determination of N(HD) requires to know precisely the b Doppler parameter

<u>Towards ζ Per :</u>

- Only HD at 1054 Ang. detected
- flat part of the curve of growth
- Re-analysis with updated H₂, f values (Abgrall et al. 1993)
 - max value: 1.1 × 10¹⁶ cm⁻²
 - instead of 5.1 × 10¹⁵ cm⁻² (Snow 1977)

Other lines of sight: same problem



Conclusion from N(HD) :

 ζ > 5 10⁻¹⁷ s⁻¹ overestimates slightly N(HD)

(with D/H = $1.5 \ 10^{-5}$)

The neutral and ionized atoms

High ζ increases the ionization degree

Efficient recombination with electrons : S⁺ does not react with H or H₂ reactive recombination dominates







Conclusion about the ζ Per line of sight

$n_{\rm el} = 100 \ {\rm cm}^{-3} \ \gamma = 2$	ζ	H_3^+	OH	HD	S
T = 60 K,	[10 ⁻¹⁷ s ⁻¹]	[cm ⁻²]	[cm ⁻²]	[cm ⁻²]	[cm ⁻²]
$N(H_2) = 4.5 \ 10^{20} \ cm^{-2}$	1	1.5 (12)	1.6 (12)	1.7 (15)	1.7 (13)
	25	3.0 (13)	4.1 (13)	1.5 (16)	2.6 (13)
 Standard value of ζ 	100	6.3 (13)	1.4 (14)	2.0 (16)	8.2 (13)
Underestimate N(H ₃ ⁺)		8.0 (14)	4.0 (13)	2.0 (15)	1.5 (13)
by a factor 50	ODS.			1.1 (16)	2.2 (13)

• ζ = 100 times the standard value and T = 60 K

Reproduce $N(H_3^+)$ but overproduce OH by a factor 4 S by a factor 6

T can be decreased to 45 K to match better OH negative impact on C, S, H_3^+

 χ can be increased to match better S molecules too much photodissociated

• $\zeta = 25 \times 10^{-17}$ s⁻¹ good compromise to fit all abundances

H₃⁺ towards the Galactic center

Observations (Oka et al. 2005)

Clouds			Т	n				
		[10 ¹⁴ cm ⁻²]						[cm ⁻³]
	(1,1)	(3,3) (2,2) (1,0) HM Tota						
-100 km s ⁻¹	7.0±0.8	4.4 ± 0.9	≤ 0.7	2.9 ± 1.0	1.4 ± 0.7	15.7 ± 1.7	270 ± 70	≤ 5 0
-50 km s ⁻¹	2.6 ± 0.5	1.6 ± 0.6	0.4 ± 0.4	1.6 ± 0.9	0.4 ± 0.2	6.6 ± 1.3	250 ± 100	≤ 100
0 km s⁻¹	4.9 ± 05	1.0 ± 07	≤ 0.7	2.4 ± 1.3	0.1 ± 0.1	8.4 ± 1.6	130 ± 100	≤ 200

Constraints : Populations $N(H_3^+)$ $N(CO, J=0) < 3 \ 10^{16}$ f = 0.5 - 1

H₃⁺ excitation

• Oka & Epp (2004) prescription for collision rates:

$$k_{JK}^{J'K'} = C_{JK}^{J'K'} \sqrt{\frac{g_{JK}}{g_{J'K'}}} \exp\left(-\frac{E_{JK} - E_{J'K'}}{2kT}\right)$$
$$C_{JK}^{J'K'} = C_{J'K'}^{JK} = C\left\{1 + \sum_{J''K''} \left(\frac{g_{J''K''}}{\sqrt{g_{JK}g_{J'K'}}}\right)^{1/2} \exp\left[-\frac{E_{J''K''} - (1/2)(E_{JK} + E_{J'K'})}{2kT}\right]\right\}^{-1}$$
$$C = 2 \times 10^{-9} \text{ cm}^3 \text{ s}^{-1} \text{ : Langevin rate constant for } H_3^+ + H_2$$

Т (K)

• Einstein coefficients from Lindsay & McCall (2001)

Small program to compute H_3^+ excitation \Rightarrow same results than Oka & Epp (2004)







Reduction of C_{langevin} :

n-T domain reproducing the observations increases higher densities & temperatures allowed

New implementations in the PDR code

1. Statistical balance of H_{3}^{+}

- H_{3^+} formed following a Boltzman distribution at T_{kin}
 - Tests: formation in specific levels
 - \checkmark no significant differences
- destruction independent of levels

2. Neutralization of atomic ions on grains



Isothermal PDR models

 χ = 10, R_X = 3

n [cm ⁻³]	ζ [10 ⁻¹⁷ s ⁻¹]	T [K]	L [pc]	f	N(CO) [cm ⁻²]	J = 1 [cm ⁻²]	N(H ₃ ⁺) [cm ⁻²]	1,1 [cm ⁻²]	2,2 [cm ⁻²]	3,3 [cm ⁻²]
	25	270	100	0.61	1.3 (17)	6.6 (16)	1.2 (15)	5.2 (14)	1.2 (13)	3.0 (14)
10	50	270	100	0.42	1.4 (17)	7.0 (16)	1.2 (15)	5.4 (14)	1.4 (13)	3.2 (14)
	100	270	100	0.23	4.4 (16)	1.7 (16)	5.7 (14)	2.5 (14)	7.4 (12)	1.5 (14)
	50	270	20	0.82	1.5 (17)	6.6 (16)	8.7 (14)	3.5 (14)	3.3 (13)	2.3 (14)
		270	31	0.83	3.4 (17)	1.3 (17)	1.8 (15)	7.3 (14)	6.9(13)	4.8(14)
		200	31	0.83	3.9 (17)	1.5 (17)	1.6 (15)	7.0 (14)	6.0 (13)	3.1 (14)
50		150	33	0.84	4.7 (17)	1.7 (17)	1.5 (15)	7.0 (14)	5.4 (13)	1.9 (14)
	100	270	22	0.70	2.4 (17)	1.0 (17)	1.7 (15)	6.7 (14)	6.9 (13)	4.5 (14)
	500	270	31	0.26	1.7 (17)	8.2 (16)	1.5 (15)	5.5 (14)	7.5 (13)	3.8 (13)
Observations		min max	15-20	0.5 1		< 3 (16)	1.4 (15) 1.7 (15)	6.2 (14) 7.8 (14)	< 7 (13)	3.5 (14) 5.3 (14)

Thermal Balance

How to have $N(H_3) = 10^{14} \text{ cm}^{-2}$ in a diffuse medium ($n_H < 50 \text{ cm}^{-3}$) at 270 K?







2 - Vortex

Local and temporal heating of the gas allowing to overcome activation thresholds

- Introduced by K. Joulain & E. Falgarone (1998)
 - ♦ formation of CH⁺

 $C^+ + H_2 \longrightarrow CH^+ + H$ $k = 1.5 \ 10^{-10} \ e^{-4640/T} \ cm^3 \ s^{-1}$

- Cecchi-Pestelini, Casu, Dalgarno (2005) Model of vortex to compute H₂ excitation
 - \clubsuit Reproduce rotationnal H₂ excitation (J > 2) in standard diffuse medium

=> test this on the H_3^+ excitation